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Chapter 1: Matter and Change (Chem in 15 minutes or less)

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CHAPTER 4 REVIEW Arrangement of Electrons in Atoms SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. d How many quantum numbers are used to describe the properties of electrons in atomic orbitals? (a) 1 (c) 3 (b) 2 (d) 4 2. a A spherical electron cloud surrounding an atomic nucleus would best represent (a) an s ...

~~4 Arrangement of Electrons in Atoms~~

CHAPTER 2 REVIEW Measurements and Calculations

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SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research team records the rainfall in inches

## ~~2 Measurements and Calculations~~

CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy

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required to remove an electron from a neutral atom is the atom's

## ~~5 The Periodic Law~~

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

## ~~3 Atoms: The Building Blocks of Matter~~

CHAPTER 9 REVIEW Stoichiometry SECTION 3

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PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of H

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CHAPTER 1 REVIEW Matter and Change SECTION 1  
SHORT ANSWER Answer the following questions in the space provided. 1. a Technological development of a chemical product often (a) lags behind basic research on the same substance. (b) does not involve chance discoveries. (c) is driven by curiosity. (d) is done for the sake of learning something new.

~~1 Matter and Change - HUBBARD'S CHEMISTRY~~  
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ARRANGEMENT OF ELECTRONS IN ATOMS The following pages contain the bulk (but not all) of the information for the chapter 4 test Focus on this content, but make sure to review class notes, activities, handouts, questions, etc

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be

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attached to the iron atom. (c) the charge on each Fe ion.

~~7 Chemical Formulas and Chemical Compounds~~  
CHAPTER 1 REVIEW Matter and Change SECTION 3  
SHORT ANSWER Answer the following questions in the space provided. 163 Chapter 11 Modern Atomic Theory Review Skills 11.1 The Mysterious Chemistry Chapter 11 Review Answers Ch 11: Holt McDougal Modern Chemistry...

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